

# CORRIGENDUM

J. Hioe, A. Karton, J. M. L. Martin,  
H. Zipse\* ..... 6861–6865

## Borane–Lewis Base Complexes as Homolytic Hydrogen Atom Donors

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In Figure 1 of the Full Paper by H. Zipse et al., the structure of compound **22** was found to have been drawn incorrectly. The corrected structure is shown in the new version of Figure 1 given below. Additionally, the data for compounds **18** and **22** were interchanged in Table 2; the values are shown in the correct order here and the compounds have been reordered in Figure 1 as a consequence. Lastly, a recalculation of the data in Tables 1 and 2 indicated a round-off error of  $0.3 \text{ kJ mol}^{-1}$  for compound **14** in Table 2. The changes do not affect the conclusions of the manuscript, and the authors apologize for any inconvenience caused by these errors.

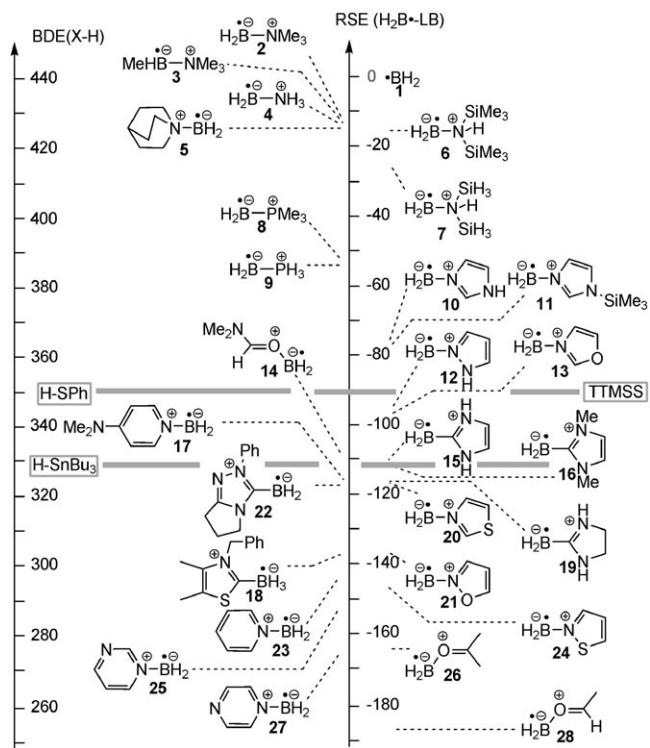


Figure 1. Radical stabilization energies (RSE, in  $\text{kJ mol}^{-1}$ ) for a variety of boryl radicals as obtained at G3(MP2)-RAD level of theory together with the BDE(B–H) values of the corresponding boranes.

Table 2. Radical stabilization energies (RSE) for a variety of boryl radicals together with the BDE(B–H) values of the corresponding boranes, the complexation energies of borane–Lewis base complexes ( $E_c(\text{H}_3\text{BLB})$ ), and complexation energies of boryl radical–Lewis base complexes ( $E_c(\text{H}_2\text{B}^\bullet\text{LB})$ ). All energies have been obtained at G3(MP2)-RAD level and are given in  $\text{kJ mol}^{-1}$ .

System	RSE( $\text{H}_2\text{B}^\bullet\text{LB}$ )	BDE( $\text{H}-\text{BH}_2-\text{LB}$ )	$E_c(\text{H}_3\text{BLB})$	$E_c(\text{H}_2\text{B}^\bullet\text{LB})$
<b>1</b>	0.0	441.1	—	—
<b>2</b>	-12.6	428.7	-158.3	-170.9
<b>3</b>	-12.8	428.3	-280.8	-139.7
<b>4</b>	-13.7	427.4	-114.1	-127.8
<b>5</b>	-14.0	427.1	-170.6	-184.6
<b>6</b>	-16.7	424.4	-99.9	-116.7
<b>7</b>	-24.6	416.5	-158.3	-113.8
<b>8</b>	-52.1	389.0	-159.9	-212.0
<b>9</b>	-54.3	386.8	-91.4	-145.6
<b>10</b>	-77.3	363.8	-136.7	-213.9
<b>11</b>	-79.4	361.7	-142.8	-222.2
<b>12</b>	-95.2	345.9	-129.0	-224.2
<b>13</b>	-97.9	343.2	-119.7	-217.6
<b>14</b>	-108.9	332.2	-84.1	-193.0
<b>15</b>	-110.7	330.4	-220.6	-331.3
<b>16</b>	-112.5	328.6	-228.0	-340.5
<b>17</b>	-116.3	324.8	-147.5	-263.8
<b>18</b>	-138.7	302.4	-216.0	-354.7
<b>19</b>	-116.9	324.2	-228.1	-345.0
<b>20</b>	-117.7	323.4	-124.4	-242.0
<b>21</b>	-137.2	303.9	-107.3	-244.5
<b>22</b>	-116.6	324.5	-211.1	-327.7
<b>23</b>	-144.3	296.8	-132.0	-276.3
<b>24</b>	-146.6	294.5	-109.2	-255.8
<b>25</b>	-152.5	288.6	-126.1	-278.6
<b>26</b>	-163.6	277.5	-73.6	-237.2
<b>27</b>	-165.8	275.3	-126.5	-292.2
<b>28</b>	-184.0	257.1	-66.1	-250.1